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Authigenic pyrite framboids in sedimentary facies of the Mount Wawel Formation (Eocene), King George Island, West Antarctica

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Abstract: Pyrite framboids occur in loose blocks of plant-bearing clastic rocks related to volcano-sedimentary succession of the Mount Wawel Formation (Eocene) in the Dragon and Wanda glaciers area at Admiralty Bay, King George Island, West Antarctica. They were investigated by means of optical and scanning electron microscopy, energy-dispersive spectroscopy, X-ray diffraction, and isotopic analysis of pyritic sulphur. The results suggest that the pyrite formed as a result of production of hydrogen sulphide by sulphate reducing bacteria in near surface sedimentary environments. Strongly negative $\delta^{34}S_{VCDT}$ values of pyrite (-30 – -25 %) support its bacterial origin. Perfect shapes of framboids resulted from their growth in the open pore space of clastic sediments. The abundance of framboids at certain sedimentary levels and the lack or negligible content of euhedral pyrite suggest pulses of high supersaturation with respect to iron monosulphides. The dominance of framboids of small sizes $(8-16 \,\mu\text{m})$ and their homogeneous distribution at these levels point to recurrent development of a laterally continuous anoxic sulphidic zone below the sediment surface. Sedimentary environments of the Mount Wawel Formation developed on islands of the young magmatic arc in the northern Antarctic Peninsula region. They embraced stagnant and flowing water masses and swamps located in valleys, depressions, and coastal areas that were covered by dense vegetation. Extensive deposition and diagenesis of plant detritus in these environments promoted anoxic conditions in the sediments, and a supply of marine and/or volcanogenic sulphate enabled its bacterial reduction, precipitation of iron monosulphides, and their transformation to pyrite framboids.

Key words: Antarctica, King George Island, Eocene, framboidal pyrite, bacterial sulphate reduction, sulphur isotopes.

Introduction

One of the most common forms of pyrite in sedimentary rocks are framboids. Framboids, as defined by Ohfuji and Rickard (2005), are microscopic spheroidal

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Fig. 1. A. Location of King George Island in Antarctica. **B**. Tectonic map of King George Island showing position of the Warszawa Block and location of plant fossil site in the Mount Wawel Formation. Map after Birkenmajer (1983).

to sub-spheroidal clusters of equidimensional and equimorphic microcrystals. This definition embraces three fundamental characteristics of framboids: (1) they are spheroidal to sub-spheroidal in form, (2) they are composed of densely packed discrete microcrystals, and (3) the microcrystals are equidimensional and equimorphic. Framboids in sedimentary facies originate due to crystallization of pyrite from amorphous or poorly ordered iron sulphides that precipitate in the form of spherical microbodies in the anoxic sulphidic environment (Berner 1970, 1984, 1985; Canfield 1989). These microspheres reflect precipitation from highly supersaturated fluids developed due to metabolic activity of sulphate reducing bacteria (Chambers and Trudinger 1979; Habicht and Canfield 1997; Wilkin and Barnes 1997a, b).

This paper presents pyrite framboids occurring in a plant-bearing volcanosedimentary succession of the Mount Wawel Formation (Eocene) that crops out in the Dragon and Wanda glaciers area at Admiralty Bay, King George Island, West Antarctica (Fig. 1). Geochemical, textural, and morphological data are presented to support bacterial origin of the framboids during authigenesis of organic-rich sediments in terrestrial environments of the northern Antarctic Peninsula region.



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Geological background

King George Island (KGI) is located in the South Shetland Islands (Fig. 1) that make an outer magmatic arc related to the subduction-related magmatism in the northern Antarctic Peninsula region (Birkenmajer 1983, 2001). On the north it borders the South Shetlands subduction trench, which is a part of the Drake Passage tectonic structure. On the south it is separated from the crustal terrane of Antarctic Peninsula by a young rift and basin of the Bransfield Strait. KGI consists of several tectonic blocks showing considerable differences in stratigraphic succession, geological ages and character of rocks (Birkenmajer *et al.* 1991). The Warszawa Block occurs in the southern part of the island, and is dominated by volcanic and volcano-sedimentary formations of Paleogene age that are classified into the Admiralty Bay, Ezcurra Inlet, and Point Hennequin groups (Birkenmajer 2003).

The Mount Wawel Formation (300 m thick) constitutes the upper part of the Point Hennequin Group that crops out along Martel Inlet on the eastern side of Admiralty Bay (Fig. 2). The formation overlies the Viéville Glacier Formation, and its upper boundary is erosional (Birkenmajer 1980). It consists of several thick andesite lava flows, and embraces three informal sedimentary units containing plant remains: the Dragon Glacier plant beds, the Wanda Glacier plant beds, and the Mount Wawel plant beds (Birkenmajer 1981; Zastawniak *et al.* 1985; Zastawniak 1998). Abundant plant fossils, cases of caddisflies and associated trace fossils (Uchman *et al.* 2008) occur in clastic rocks distributed mainly in the form of loose blocks on lateral and marginal moraines of the Dragon and Wanda glaciers. A Middle Eocene to Late Oligocene age for the Point Hennequin Group was suggested on the basis of radiometric datings (Pankhurst and Smellie 1983; Smellie *et al.* 1984; Birkenmajer 1989, 2003). However, a new series of K-Ar datings of the



Fig. 2. Geological panorama of the southern coast of Martel Inlet in Admiralty Bay (after Birkenmajer 1983) showing location of samples analyzed in this paper.







Fig. 3. Framboidal pyrite in sedimentary facies of the Mount Wawel Formation (I). **A**, **B**. SEM images of polyframboidal aggregates. **C**, **D**. TLM and RLM images of polyframboidal aggregates (normal light). Framboids occurring in a mouldic pore after partial decomposition of a plant debris are shown in C.

Mount Wawel Formation points to its exclusive Eocene age (Z. Pécskay, personal communication). Loose blocks of plant-bearing clastic rocks in the Dragon and Wanda moraines have not been as yet precisely dated (Birkenmajer 1980, 1981, 1989; Zastawniak 1981; Zastawniak *et al.* 1985; Birkenmajer and Zastawniak 1989). Similar plant beds found in the Fildes Formation on Fildes Peninsula (Li 1994; Zhou 1994) and in the Point Thomas Formation in Ezcurra Inlet (KGI) have been dated as Eocene (Poole *et al.* 2001; Hunt and Poole 2003; Tatur *et al.* 2009).

Plant remains and the associated authigenic pyrite occur in fine-grained rocks, mostly in grey and greyish-brown mudstone, siltstone and claystone, which contain small-sized volcanic bombs (up to a few centimetres in size) and scattered finer volcanogenic clasts. The clastic material is dominated by recrystallized brown and green volcanic glass. The sediment reveals stratification emphasized by vertical changes in the grain size as well as by horizontal alignment of plant remains in fine-grained intervals. Sedimentary structures suggest deposition in stagnant and flowing water environments, with recurrent pulses of increased supply of plant ma-



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Fig. 4. Framboidal pyrite in sedimentary facies of the Mount Wawel Formation (II). A–C. TLM and RLM images of polyframboidal aggregates (normal light). Ordered (*o*) and disordered (*d*) framboids are shown in A. D. BSE image of polyframboidal aggregate.

terial. This material is represented by fragments of Gymnosperm shoots and Angiosperm leafs, various seeds and fruit scales as well as undefined plant detritus (Zastawniak *et al.* 1985). The Angiosperms are dominated by the genus *Nothofagus*, and the Gymnosperms by Podocarpaceae. This plant community resembles evergreen or deciduous forests, similar to the ones in the present-day Patagonia, New Zealand (Zastawniak 1981; Zastawniak *et al.* 1985; Birkenmajer and Zastawniak 1989), and south-east Australia (Zastawniak 1981). The forests covered slopes and valleys of mountainous landscape in the magmatic arc, providing plant material to the fresh-water and nearshore sedimentary environments (Poole *et al.* 2001; Hunt and Poole 2003).

Materials and methods

The material analyzed in this paper was collected at three locations along the outcrop belt of the Mount Wawel Formation: (1) the lateral and marginal moraines









Fig. 5. Pyritic structures associated with framboids in sedimentary facies of the Mount Wawel Formation. A. Pyritized plant debris in a plant-bearing horizon. TLM and RLM image, normal light. **B–D**. Pyritic microcaplsules showing empty interiors and apertures (*arrows*) open towards the pore space. SEM images.

of the Wanda Glacier, (2) the lateral moraines of the Dragon Glacier, and (3) the marginal moraines of the Dragon Glacier (Fig. 2). At these locations, large accumulations of loose blocks up to a few metres in size contain horizons and layers enriched in plant detritus and authigenic pyrite.

Pyrite was identified on the basis of petrographic observations under transmitted and reflected light microscopy (TLM and RLM), and scanning electron microscopy (SEM). The three-dimensional morphology of pyrite was studied on broken rock surfaces that were subjected to etching using hydrofluoric acid. After petrographic analysis in transmitted and reflected light, thin sections were coated with carbon and analyzed under SEM. A JEOL JXA 840A scanning electron microscope operating at a 15 kV accelerating voltage was used. Quantitative EDS analyses of pyrite were obtained using the same microscope, equipped with a THERMO NORAN VANTAGE EDS system. Operating conditions were a 15 kV acceleration voltage, 1 to 5 mm beam diameter, and 100 s counting time. Detection limits of the analyzed elements (S, Fe, Co, Ni, Cu, Zn) were better than 0.05 wt. %.





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Fig. 6. Details of pyritic framboids in sedimentary facies of the Mount Wawel Formation (I). A. Example of ordered structure. B–D. Examples of disordered structure. A rim of elongated pyrite crystals (r) coating disordered framboid is shown in B. A–D SEM images.

Back-scattered electron images (BSE) were used to precisely locate analytical points on thin sections.

For the X-ray diffraction (XRD) and sulphur isotope (SI) analyses, the rock samples were crushed and powdered to the fraction 5–10 µm. Twenty and three samples containing framboidal pyrite have been selected for XRD and SI analyses, respectively. X-ray diffraction patterns were recorded on a SIGMA 2070 diffractometer using a curved position sensitive detector in the range 2–120° 20 with CoKa radiation and 20 hour analysis time. DIFFRACTIONEL software v. 03/93 was used to process the obtained data. Isotopic composition of pyritic sulphur was analyzed in 1 to 5 g of sample. The sulphur extraction procedure was based on the method presented in Kolthoff and Sandell (1952), Canfield *et al.* (1986), Tuttle *et al.* (1986), and Bates *et al.* (1993). The samples were treated by reaction with a 1.0 M solution of chromium chloride (CrCl₂) and 12 N HCl in proportion 2:1. Then the mixtures were boiled under an Ar flow. This procedure converted pyrite (FeS₂) to hydrogen sulphide (H₂S). A flow of Ar carried H₂S through the apparatus, first through the buffer solution of 80% 1 M CH₃COOH + 20% 1 M CH₃COON at pH4









 Fig. 7. Details of pyritic framboids in sedimentary facies of the Mount Wawel Formation (II). A–C.
Framboids showing interlocking mosaic of recrystallized (cubic and/or octahedral) pyrite microcrystals. D. Pyrite rims (*r*) coating ordered framboids. A–D SEM images.

to remove chlorine compounds, and then to a vessel where H_2S reacted with AgNO₃ to form Ag₂S. Ag₂S was collected by filtration, dried, and weighed. It was isotopically analyzed using a MAT 253 spectrometer equipped with a FLASH EA device in continous flow mode. Sulphur isotope ratios are reported as $\delta^{34}S$ notations relative to the Vienna Canon Diablo Troilite standard (VCDT).

All samples analyzed in this paper are housed at the Institute of Geological Sciences, Polish Academy of Sciences in Warszawa.

Results

In clastic rocks of the Mount Wawel Formation, pyrite occurs predominantly in the form of framboids that concentrate at discrete horizons enriched in plant detritus (Figs 3, 4). The framboids are found to have developed in open pore space, where they form polyframboidal seams and lenses up to a few mm in length. These pyritic bodies occur close to pyritized plant fragments, the latter showing pre-





Fig. 8. Histogram showing the frequency of framboid diameter in sedimentary facies of the Mount Wawel Formation.

snown in Fig. 11.			
No. of analysis	Pyritic S (mol %)	Pyritic Fe (mol %)	
1	64.73	34.34	
2	64.78	34.60	
3	64.41	35.01	
4	64.43	34.32	
5	65.19	34.58	
6	65.05	34.83	
7	64.82	34.90	
8	64.83	34.49	
9	64.02	35.29	
10	64.46	35.22	
11	64.06	35.57	
12	62.80	36.71	
13	65.21	34.68	
14	65.56	34.14	
15	63.62	35.60	
Min. value	62.80	34.14	
Max. value	65.56	36.71	
Average	64.53	34.95	

Molar composition of sulphur and iron in framboidal pyrite. Location of analytical points is shown in Fig. 11.

Table 1





Fig. 9. X-ray diffraction (XRD) pattern of plant-bearing mudstone enriched in framboidal pyrite. Numbers in brackets indicate *hkl* reflections of pyrite. Dragon Glacier marginal moraine; sample H-1.



Fig. 10. EDS spectrum of a pyrite framboid. Wanda Glacier marginal moraine; sample S-2.

served anatomic details without any compactional deformation (Fig. 5A). Associated with pyrite framboids occur pyritic microcapsules (8–12 μ m), often showing empty interiors and apertures open towards the pore space (Fig. 5B–D).





Fig. 11. BSE image of polyframboidal pyrite showing location of EDS point analyses listed in Table 1. Dragon Glacier marginal moraine; sample S-2.

The framboids are spheroidal to sub-spheroidal in form; however, some framboids reveal slightly ellipsoidal shapes (Figs 3D, 4A–D). They are composed of densely packed discrete microcrystals. The microcrystals reveal equidimensional and equimorphic structure. Based on SEM observations, two groups of the framboid structure can be discerned: ordered and disordered. The ordered structure consists of microcrystals that show almost uniform morphological alignments (Figs 4A, 6A), whereas the disordered structure is made up of chaotically distributed microcrystals (Figs 4A, 6B–D). The ordered forms are more frequently observed. Framboids showing overgrowths with layers of densely packed, elongated pyrite crystals are also observed (Figs 6B, 7D). Some framboids show diagenetic recrystallization that is manifested by replacements by larger cubic and/or octahedral crystals (Fig. 7A–C).

Statistical analysis of the framboid diameters measured in thin sections (more than 1,400 measurements) shows that the framboids vary in size from < 1 μ m up to 40 μ m, with more than 50% falling in a range between 8 μ m and 16 μ m (Fig. 8). Large scattered framboids (30–40 μ m) were observed in all the investigated samples.

The results of XRD analysis show the presence of pyrite, without any detectable admixture of other sulphide minerals (Fig. 9). The results of EDS analysis show that most of the framboids are composed of pure pyrite, with the atomic Fe:S ratio close to 1:2 (Table 1, Figs 10, 11). Minor amounts of iron oxides have been detected in external parts of some framboids, possibly resulting from epigenetic oxidation of pyrite. In some framboids, trace amounts of Co, Ni, Cu, and Zn were detected.





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The δ^{34} S values of framboidal pyrite in three analyzed samples fall in a narrow range between -30 % and -25 % (Table 2).

Table 2

Results of sulphur isotopic	analysis ($\delta^{34}S_{vCDT}$	%) of framboidal pyrite
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Sample	δ ³⁴ S (%)
H-1	-25.4
S-2	-29.2
F-2	-30.1

Discussion

Sulphur may be present in sedimentary facies in the form of iron monosulphides (mackinawite, pyrrohotite or an amorphous phase), elemental sulphur, sulphates, organically-bound sulphur, and most commonly as pyrite (Canfield et al. 1986; Gorjan et al. 2003). The content and speciation of sulphur in sediment is related to the dominant type of diagenetic environment, in which various sulphur compounds are deposited as a result of its oxidation and/or reduction reactions (Berner 1974, 1984; Goldhaber and Kaplan 1974). Therefore, sulphur is considered a tracer of biogeochemical processes in sedimentary environments, and used as a proxy of paleoredox conditions in the geological record (e.g. Jones and Manning 1994; Lyons et al. 2003). The prevailing occurrence of authigenic pyrite is thought to reflect dominant anoxic sulphidic conditions (Raiswell et al. 1988). However, this pyrite can have diverse morphologies. Two of the most commonly observed are framboids and micron-sized euhedral (microgranular) crystals (Wang and Morse 1996; Taylor and Macquaker 2000). It has been argued that euhedral pyrite precipitates from porewaters oversaturated with respect to pyrite, but undersaturated with respect to iron monosulphides (Wang and Morse 1996). In contrast, pyrite framboids are commonly believed to have formed through iron monosulphide intermediates, from porewaters supersaturated with respect to both pyrite and iron monosulphides (Sweeney and Kaplan 1973; Morse et al. 1987; Roberts and Turner 1993; Taylor and Macquaker 2000).

In the study material, sulphur is predominantly bound in the form of pyrite framboids. The absence of euhedral pyrite indicates that the early phase of framboidal pyrite formation in sediments of the Mount Wawel Formation was associated with high supersaturation conditions with respect to iron monosulphides. Framboids are abundant throughout all the analyzed pyritization horizons and layers, where they are associated with pyritized detritus of higher plants showing well-preserved anatomic features. Results of experimental precipitation of pyrite within cells of higher plants at ambient temperatures were described in detail by Rickard et al. (2007). The presence of pyrite in that form is indicative of point concentration of ions participated in the chemical reaction (HS⁻ and Fe²⁺). High con-



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centration of these ions leads to abrupt precipitation of iron monosulphides showing microglobular fabric. The abundance of well-developed framboids indicates that their formation was a relatively rapid process associated with the presence of highly reactive organic matter (Billon *et al.* 2007). The nature of the preserved plant detritus in sediments of the Mount Wawel Formation suggests pulses of enhanced deposition of fresh organic matter from the immediate neighbourhood, which is consistent with paleoenvironmental reconstructions of the plant-bearing strata (Zastawniak 1981; Zastawniak *et al.* 1985; Birkenmajer 1989; Poole *et al.* 2001; Hunt and Poole 2003).

Investigations of modern and ancient sedimentary environments suggest that the distribution of sizes of framboids is a reliable indicator of the extent of anoxic sulphidic conditions (Wilkin et al. 1996). Under euxinic conditions, syngenetic framboids form directly in the water column, reaching maximum diameter of approximately 5 µm before falling out of the water column. They do not continue to grow after the accumulation in bottom sediments since they are no longer close to the redox boundary. In contrast, in sedimentary environments underlying dysoxic or oxic water column the framboids are larger and vary in size. As reported by Marynowski et al. (2007), a further increase of size may be caused only by diagenetic changes, such as formation of euhedral pyrite crystals. Wilkin et al. (1996), Wignall and Newton (1998), Wignall et al. (2010), Bond et al. (2010) suggest that larger framboids form within the sediments below an oxic and/or dysoxic water column. They observe that in sediments accumulated in euxinic environments the mean diameter of framboids is small $(2.7-3.2 \,\mu\text{m})$, and that framboids bigger than 10 µm do not occur. The mean diameter increases under conditions of incipient dysoxia, and there may appear single framboids of more than 10 µm in size. On the other hand, under strictly dysoxic conditions a rapid increase of framboid diameter is observed.

In all the samples investigated here, the framboids are characterized by a wide range of diameters from < 1 μ m up to 40 μ m, which confirms the development of anoxic sulphidic conditions within the sediment in well-aerated terrestrial environments. The abundance of framboids of small sizes (8–16 μ m) in the plant-bearing horizons suggests that these conditions developed repeatedly, and led to rapid pulses of iron monosulphide precipitation associated with intense decomposition of organic matter. All the framboids are found to occur in the primary pore space without any traces of compressional deformation (*e.g.* flattening), which suggests that their pyritic structure have formed before any compaction of the sediment, *i.e.* very early during authigenesis. Incorporation of elemental sulphur originated either as a result of hydrogen sulphide oxidation or sulphate reduction was a likely scenario for transformation of monosulphides into pyrite (Berner 1970; Middelburg 1991; Billon *et al.* 2007).

It is generally accepted that the formation of framboidal pyrite in sedimentary environments is closely associated with activity of sulphate reducing bacte-





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ria. Bacterial sulphate reduction is a low temperature process; and it is common in diagenetic settings from 0°C up to about 60–80°C at maximum (Machel 2001; Canfield *et al.* 2006). Above this temperature range, almost all sulphate reducing bacteria cease to metabolize. Pyritic horizons observed in the Mount Wawel Formation represent recurrent events of rapid precipitation that followed deposition of plant detritus and intense decomposition of organic matter. Associated with the framboids occur pyritic microcapsules, which are interpreted to represent authigenic mineral encrustations on microbial cells showing metabolic activity (Krajewski *et al.* 1994). The pyritized microbial ultrastructures document an abundance of microbial communities in the sediment, and confirm the formation of pyrite framboids at sedimentary temperatures. However, volcanic bombs found in some plant-bearing horizons suggest deposition in close proximity to volcanic activity.

It is well known that the isotopic composition of sulphur in sedimentary pyrite corresponds to the isotopic composition of the dissolved sulphide in the precipitating medium (Butler et al. 2004). During bacterial sulphate reduction, the light sulphur isotope $({}^{32}S)$ is preferentially metabolized causing the generated hydrogen sulphide to be isotopically light compared to the source sulphate (Kaplan and Rittenberg 1963). Laboratory and field studies show that a common sulphur isotope fractionation during this process ranges between 20 % and 40 % (Butler et al. 2004; Mangalo *et al.* 2007). The δ^{34} S values of pyrite in the Mount Wawel Formation fall in a narrow range from -30 % to -25 %, suggesting that the source sulphate had isotopic composition roughly between 0 % and +20 %. Sulphate in the Eocene sedimentary environments of KGI could originate either from volcanic eruptions or from sea spray, or represent a mixture of the two sources. The $\delta^{34}S$ values for volcanic andesite-type rocks range from 0 % to +20 % (Torssander 1992; Marini et al. 2002; Mazot et al. 2007). The isotopic composition of sulphur in marine sulphate is close to +22 % (Thode 1991), and similar values are expected to have dominated the Paleogene marine water (Ayora et al. 1995). It seems therefore likely that the sulphate subjected to bacterial reduction in sediments of the Mount Wawel Formation was of marine or mixed marine-volcanic origin.

Conclusions

(1) The plant-bearing clastic facies of the Mount Wawel Formation (Eocene), King George Island, West Antarctica is enriched in authigenic pyrite, which occurs mostly in the form of framboids associated with pyritized plant debris.

(2) The pyrite framboids originated in the nearsurface environment of clastic sediments, where enhanced degradation of organic matter contributed to the development of anoxic sulphidic conditions. Isotopic composition of pyritic sulphur indicates that hydrogen sulphide generated in this environment was a result of activity of



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sulphate reducing bacteria. Textural and chemical features of framboids suggest that their formation involved intermediate precipitates of amorphous or poorly ordered iron monosulphides, followed by crystallization of the mineral phase.

(3) The clastic facies of the Mount Wawel Formation records sedimentation in local depressions and coastal areas on a young magmatic island arc, with recurrent events of enhanced supply of plant detritus from forest-covered mountainous land-scape. The abundance of authigenic pyrite in this facies suggests that there was a supply of marine or mixed marine-volcanic sulphate to sedimentary environments of the arc islands.

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